## **CHEMISTRY**

**Lewis Structure & Dipoles Worksheet** 

<b>NAME</b>		
-------------	--	--

- Draw Lewis structures AND draw dipoles for the following substances in the space below.
- Tell the SHAPE of each molecule.
- Predict the TYPE of molecule whether polar or nonpolar.

<b>IMPORTANT:</b> (Cl is chlorine	- there is no iodine on this sheet.)
-----------------------------------	--------------------------------------

A) CIF SHAPE: \_\_\_\_\_ TYPE: \_\_\_\_

B) SiF<sub>4</sub> SHAPE: \_\_\_\_\_ TYPE: \_\_\_\_

C) Cl<sub>2</sub>O SHAPE: \_\_\_\_\_ TYPE: \_\_\_\_

D) CH<sub>2</sub>Br<sub>2</sub> SHAPE: \_\_\_\_\_ TYPE: \_\_\_\_

E) CS<sub>2</sub> SHAPE: \_\_\_\_\_ TYPE: \_\_\_\_

F)	$BCl_3$	SHAPE:	TYPE:
G)	C <sub>2</sub> Cl <sub>4</sub>	SHAPE:	TYPE:
H)	$SO_3$	SHAPE:	TYPE:
I)	$C_2F_2$	SHAPE:	TYPE:
J)	$N_2O_2$	SHAPE:	TYPE: